

CHEM 2302

Fall 2026



Instructor:

Professor Jeff Smith

Office: Steacie 225

Phone: (613) 520-2600 ext.2408

E-mail: jeff.smith@carleton.ca

Lectures:

Mondays and Wednesdays, 2:35pm-3:55pm

In person in Tory Building room 208

Textbook:

Quantitative Chemical Analysis by Daniel C. Harris, 11th Ed., Freeman, 2026. Available on the Macmillan website but note that you get the e-book when you purchase Achieve (\$129.99).

<https://www.macmillanlearning.com/college/ca/product/Quantitative-Chemical-Analysis/p/1319164307>

Course website:

Brightspace will be used. Please visit

<https://brightspace.carleton.ca> for course-related materials.

Office hours:

By appointment only (email me to find a time).

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Course description: Introduction to quality assurance measures, calibration strategies and the fundamentals of solution-based analytical measurement processes. Qualitative and quantitative analysis using potentiometric and electrolysis techniques including ion selective electrodes, coulometry, amperometry and voltammetry. Redox, acid/base and EDTA titrations in the context of various buffer systems.

Preclusion:

Additional credit for CHEM 2300

Prerequisites:

CHEM 1006 with a minimum grade of B-, or CHEM 1002 or CHEM 1101 and (MATH 1007 or MATH 1004)

Lectures three hours a week, laboratory three hours a week

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Homework Management System:

This year a homework management platform will be bundled with the QCA11e textbook called Achieve. It will allow you to work through problems related to the material we are learning to solidify concepts in your mind and help you understand how to relate the theoretical knowledge you are learning to practical situations. The material on Achieve will resemble the types of questions that you will see on Midterms and the Final Exam.

There will be a component of your final grade attributed to your interactions with the Achieve system.

You can access Achieve through this course in Brightspace.

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Homework Management System:

Please note that Achieve is a 3rd party tool that is integrated into Brightspace and has been vetted by Carleton University for privacy and security. It is not directly licensed or supported by the University. However, I have determined that it is the best tool to meet the learning outcomes in this course. If you have questions or need support with this tool, please contact me. If you have accessibility issues using this tool, please let me know and I will work with the Paul Menton Centre to find solutions for you.

Clickers:

iClicker will be used for class polling every week for interactive and educational purposes. Cell phones, smart phones and WiFi connected laptops may be used to interact with these polls via text messaging and online answering. The use of iClicker is free with purchase of Achieve, class participation is encouraged, but not mandatory. You must link your iClicker account with this course within Achieve the first time you use it.

Laboratory:

Section A1 – Thursday 8:35-11:25am – SC 305

Section A3 – Thursday 1:35-4:25pm – SC 305

Section A4 – Friday 8:35-11:25am – SC 305

Section A5 – Wednesday 8:35-11:25am – SC 305

WHMIS:

All students registered in this course must successfully complete online training in the **Workplace Hazardous Materials Information System** (WHMIS) before being allowed to enter the laboratory. The training course takes about 2 hours to complete. If you have done the training before, there is no need to repeat it. If you have not completed the training, email the CHEM 2302 Laboratory Coordinator Graham Galway at grahamgalway@cunet.carleton.ca.

Laboratory Coordinator:

Graham Galway – grahamgalway@cunet.carleton.ca

Teaching assistants:

- Sarah Burry sarahburry@cmail.carleton.ca
- Sarah Larose sarahlarose3@cmail.carleton.ca
- Jocelyn Menard jocelynmenard@cmail.carleton.ca
- Grace Tseng gracetseng@cmail.carleton.ca

Laboratory details:

(a) Laboratories are scheduled to begin in the week of September 14th. The first lab will be an introductory session that is **mandatory to attend** to learn about the format of labs in CHEM 2302 as well as be trained on basic lab skills that will pertain to the rest of the labs in this course – please make sure you attend!

(b) The lab timetable is printed on p. 5 in the lab manual; students should prepare for each lab accordingly. The new lab manual will be available electronically (via Brightspace) and careful notes on each lab should be made by each student.

Laboratory details (cont.):

(c) The wearing of safety glasses and lab coat is MANDATORY at all times. They can be purchased from the Science Stores. Contact lenses should NOT be worn in the lab. Prescription glasses with safety lenses are permissible; if the glasses are small, goggles will have to be used. **See the safety rules on page 6 of the lab manual.**

(d) Due to the timing and equipment limitations, it is impossible to coordinate the labs and lectures. The lab manual contains some general information in the Introduction section of each experiment. Students may refer to the textbook for more theory of the technique and method.

(e) Satisfactory completion of the lab is required to pass the course. This includes performing all the experiments as scheduled, recording the output signals in the lab, analysing the data at home, and submitting your results on time in either an analysis report form (pages 16-18 in the lab manual), a short report (page 19 in the lab manual) or a formal report (pages 20-21 of lab manual).

Late deductions: 10% per business day. See lab manual for more specific details.

Laboratory details (cont.):

CHEM 2302 Laboratory Primer Videos!

These short (2-3 min) videos will give you an overview of what to expect in the labs for CHEM 2302 to help you prepare and manage your expectations going into each lab. **They are not a substitute for reading the lab manual!** They are to complement the lab manual and help you understand what you will be doing as you read through the lab manual (and do the pre-lab) prior to every lab. The goal of these videos is to help you hit the ground running for each lab to ensure success and efficient time management.

Experiment 1 – Ion Chromatography of Anions in Drinking Water with Conductivity Detection

Experiment 2 – Determination of Fluoride in Drinking Water by Ion Selective Electrode Potentiometry

Experiment 3 – Determination of Lead, Cadmium and Copper in Drinking Water by Differential Pulse Anodic Stripping Voltammetry

Experiment 4 – Determination of Copper in Drinking Water by Atomic Absorption Spectroscopy

Experiment 5 – Gas Chromatography of Polycyclic Aromatic Hydrocarbons with Flame Ionization Detection

Experiment 6 – Monitoring Carbonyls by 2,4-Dinitrophenylhydrazine Derivatization and Reversed-Phase High-Performance Liquid Chromatography

Experiment 7 – Determination of Vitamin C using Oxidation-Reduction Titrimetry

<https://www.youtube.com/watch?v=er0xs2834l8&list=PLABFzZMgnynH1oojzoxCL72sMJvJ1qU5g&pp=gAQBiAQB>

Assignments: Questions and problems will be assigned through the Achieve Homework Platform to help solidify your understanding of the class material. **Interaction with Achieve will be graded.**

Grading:

The final grade for Chemistry 2302 will be based on the following:

Laboratory component:

1) Formal reports (2 @ 7%)	14.0%
2) Short report (1 @ 2.1%)	2.1%
3) Analysis reports (4 @ 1.4%)	5.6%
4) Lab notebook (post-lab)	2.8%
5) Lab assignment	5.25%
6) Teaching assistant's evaluation of correct laboratory techniques, manipulative skills, and pre-lab preparation	3.5%
7) Online pre-lab quizzes on Brightspace	1.75%
Laboratory subtotal:	35.0%

Achieve:	5.0%
Midterm #1:	12.5%
Midterm #2:	12.5%
Final Exam:	35.0%
Total:	100.0%

(Bell curving will not be used on laboratory, midterm, exam or final marks)

Academic Accommodations and Regulations

Carleton is committed to providing academic accessibility for all individuals. You may need special arrangements to meet your academic obligations during the term. The accommodation request processes are outlined on the Academic Accommodations website (<https://students.carleton.ca/course-outline/>).

Statement on Chat GPT/Generative AI usage

As our understanding of the uses of AI and its relationship to student work and academic integrity continue to evolve, the use of AI may become permissible in this course in the future. However, at the present moment the use of AI in CHEM 2302 is not allowed. Students who are suspected of using AI (Chat GPT/Generative AI) will have their work forwarded to the Office of the Dean of Science for an investigation into academic misconduct.

Statement on Academic Integrity

Students are expected to uphold the values of academic integrity, which include fairness, honesty, trust, and responsibility. Examples of actions that compromise these values include but are not limited to plagiarism, accessing unauthorized sites for assignments or tests, unauthorized collaboration on assignments or exams, and using artificial intelligence tools such as ChatGPT when your assessment instructions say it is not permitted.

Misconduct in scholarly activity will not be tolerated and will result in consequences as outlined in [Carleton University's Academic Integrity Policy](#). A list of standard sanctions in the Faculty of Science can be found [here](#).

Additional details about this process can be found on [the Faculty of Science Academic Integrity website](#). Students are expected to familiarize themselves with and abide by [Carleton University's Academic Integrity Policy](#).

Student Rights & Responsibilities

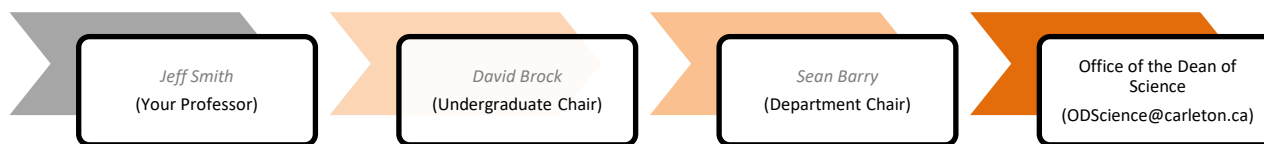
Students are expected to act responsibly and engage respectfully with other students and members of the Carleton and the broader community. See the [7 Rights and Responsibilities Policy](#) for details regarding the expectations of non-academic behaviour of students. Those who participate with another student in the commission of an infraction of this Policy will also be held liable for their actions.

Mental Health and Wellness:

As a student you may experience a range of mental health challenges that significantly impact your academic success and overall well-being. If you need help, please speak to someone. There are numerous resources available both on- and off-campus to support you. For more information, please consult <https://wellness.carleton.ca/>.

Student Concerns

If a concern arises regarding this course, **your first point of contact is me**: Email me or visit my office and I will do my best to address your concern. If I am unable to address your concern, the next points of contact are (in this order):



Note: You can also bring your concerns to [Ombuds services](#).

Assistance for Students

Writing and Learning Support: <https://carleton.ca/csas/support/>

Peer Assisted Study Sessions (PASS): <https://carleton.ca/csas/pass/>

Math Tutorial Centre: <https://carleton.ca/math/math-tutorial-centre/>

Science Student Success Centre: <https://sssc.carleton.ca/>

Academic Accommodation for Students with Religious Obligations:

Carleton University accommodates students who, due to religious obligation, must miss an examination, test, assignment deadline, laboratory, or other compulsory event. The University has a Senate-approved policy on religious accommodation that forms part of its Human Rights Policy, available at:

www.carleton.ca/equity. Students requesting academic accommodation on the basis of religious observance should make a formal, written request to their instructors for alternate dates and/or means of satisfying academic requirements. Such requests should be made during the first two weeks of class, or as soon as possible after the need for accommodation is known to exist, but no later than two weeks before the compulsory academic event. Accommodation is to be worked out directly and on an individual basis between the student and the instructor(s) involved. Instructors will make accommodations in a way that avoids academic disadvantage to the student. Students or instructors who have questions or want to confirm accommodation eligibility of a religious event or practice may refer to the Equity Services website for a list of holy days and Carleton's Academic Accommodation policies, or may contact an Equity Services Advisor in the Equity Services Department for assistance. Students unable to reach a satisfactory arrangement with their instructor(s) should contact the Director of Equity Services. Instructors who have questions or wish to verify the nature of the religious event or practice involved should also contact this officer. For more details visit the Equity Services website

http://www.carleton.ca/equity/accommodation/student_guide.htm

<http://interfaith-calendar.org/>

Academic Accommodation for Pregnant Students:

Pregnant students requiring academic accommodations are encouraged to contact an Equity Advisor in Equity Services to complete a letter of accommodation. The student must then make an appointment to discuss her needs with the instructor at least two weeks prior to the first academic event in which it is anticipated the accommodation will be required. For more details visit the Equity Services website

http://www.carleton.ca/equity/accommodation/student_guide.htm.

Plagiarism:

Plagiarism is presenting, whether intentional or not, the ideas, expression of ideas or work of others as one's own. Plagiarism includes reproducing or paraphrasing portions of someone else's published or unpublished material, regardless of the source, and presenting these as one's own without proper citation or reference to the original source. Examples of sources from which the ideas, expressions of ideas or works of others may be drawn from include but are not limited to: books, articles, papers, literary compositions and phrases, performance compositions, chemical compounds, art works, laboratory reports, research results, calculations and the results of calculations, diagrams, constructions, computer reports, computer code/software, and material on the Internet.

Examples of plagiarism include, but are not limited to:

- submitting a take-home examination, essay, laboratory report or other assignment written, in whole or in part, by someone else;
- using ideas or direct, verbatim quotations, paraphrased material, algorithms, formulae, scientific or mathematical concepts, or ideas without appropriate acknowledgment in any academic assignment;
- using another's data or research findings;
- submitting a computer program developed in whole or in part by someone else, with or without modifications, as one's own;
- failing to acknowledge sources through the use of proper citations when using another's works and/or failing to use quotation marks.

For more information, see: <http://science.carleton.ca/students/academic-integrity/>

Academic Integrity: Consequences of Academic Misconduct

In the Faculty of Science, sanctions for misconduct under the Academic Integrity Policy of Carleton University will normally be applied as follows:

- A first offence will result in a grade of zero on the work(s) associated with the misconduct, and a deduction of up to three (3) grade points from the final course grade (e.g. a grade of B could be reduced to a C).
- Any subsequent offences will result in increasingly severe sanctions ranging from:
 - A final grade of F in the course.
 - Suspension from studies for up to three (3) semesters.
 - Expulsion of enrollment from the University.

Note: These sanctions are provided here as guidelines only; more severe sanctions may be applied as appropriate (e.g., in the case of cheating on an examination).

CHEM 2302 Midterm and Final Exam Details

Here is how the midterm and final exam component of CHEM 2302 is going to work:

- There will be 2 midterm exams, one on October 21st and one on November 18th
 - *These midterms will be designed to take 60 minutes to complete but you will have the whole class time (80 minutes) to finish*
 - *The midterms will be held in person during normal class time (or in the MEC if you have approved accommodation(s) through the PMC)*
 - *Make up midterms will not be offered, if a midterm is missed, the weight of the other midterm and final exam will be adjusted to compensate for the removal of the midterm. If both midterms are missed, your final exam will be worth 60% (not recommended!!)*
- There will be a 3-hour final exam that will take place in December 2026
 - *The exam will be designed to take 120 minutes to complete but you will have 180 minutes to finish*
 - *The exam will be in person and administered by exam services (e.g. in the field house, or in the MEC if you have approved accommodation(s) through the PMC)*

CHEM 2302 Learning outcomes

- Understand and use analytical terminology
- Conduct quantitative analysis using a calibration curve
- Understand the quality assurance process
- Understand and know how to use chromatographic separation techniques
- Appreciate chemical equilibrium as it relates to analytical chemistry
- Understand how electrochemistry may be used for quantitative analysis
- Understand how buffered solutions are used in analytical chemistry
- Appreciate the many ways analytical chemistry may be used in the world
- Gain practical experience in the use of analytical instrumentation
- Be able to collect and analyze data and report results
- Enhance scientific communication skills in written form
- Improve good laboratory practice (GLP) skills

Tentative Lecture Schedule**

#	<u>Lecture</u>	<u>Topics</u>	<u>Chapters</u> (N = notes)
1	Sept 9 th , 2026	• Presentation / discussion of syllabus • Tips for success • Introduction to the analytical process • Sample preparation, liquid and solid-phase extraction	N/A N/A Ch. 0 Ch. 28-2, 28-3
2	Sept 14 th , 2026	• Brief tutorial on measurements • Brief tutorial on analytical chemistry laboratory tools • Brief tutorial on experimental error • Brief tutorial on statistics • Error calculations	Ch. 1 Ch. 2 Ch. 3 Ch. 4-1, 4-3 Ch. 4-8
3	Sept 16 th , 2026	• Quality assurance and calibration methods	Ch. 5
4	Sept 21 st , 2026	• Introduction to gas chromatography	Ch. 24
5	Sept 23 rd , 2026	• Introduction to liquid chromatography	Ch. 23-2
6	Sept 28 th , 2026	• Introduction to atomic spectroscopy	Ch. 21

** – Lecture and test / Midterm schedule is subject to change at the discretion of Professor Smith

Tentative Lecture Schedule**

#	<u>Lecture</u>	<u>Topics</u>	<u>Chapters</u> (N = notes)
7	Sept 30 th , 2026	• Chemical equilibrium constant and thermodynamics • Solubility product	Ch. 6-1, 6-2 Ch. 6-3
8	Oct 5 th , 2026	• Acids and bases and pH	Ch. 6-5, 6-6, 6-7
9	Oct 7 th , 2026	• Titrations	Ch. 7
	Oct 12 th , 2026	• <i>THANKSGIVING MONDAY</i>	N/A
10	Oct 14 th , 2026	• Ionic strength, activity coefficients and true pH	Ch. 8-1, 8-2, 8-3
11	Oct 19 th , 2026	• <i>Buffer time / optional review session for Midterm #1</i> • Fundamentals of electrochemistry • Galvanic cells • Standard potentials	N/A Ch. 14-1 Ch. 14-2 Ch. 14-3
	<u>Oct 21st, 2026</u>	• Midterm #1 on material covered until Oct 14th <i>(Test will be designed to take 60 minutes to complete; however, 80 minutes will be allocated)</i>	Ch. 0, 28-2, 28-3, 5, 6, 7, 8, 21, 23-2, 24, N

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Tentative Lecture Schedule**

#	<u>Lecture</u>	<u>Topics</u>	<u>Chapters</u> (N = notes)
	Oct 26 th – 30 th , 2026	• <i>READING WEEK</i>	N/A
12	Nov 2 nd , 2026	• Nernst Equation • E° and the Equilibrium Constant	Ch. 14-4 Ch. 14-5
13	Nov 4 th , 2026	• Electrodes and potentiometry • Reference electrodes • Indicator electrodes • pH measurement • Ion-selective electrodes	Ch. 15 Ch. 15-1 Ch. 15-2 Ch. 15-5 Ch. 15-4, -6, -7
14	Nov 9 th , 2026	• Redox titrations	Ch. 16
15	Nov 11 th , 2026	• Fundamentals of electrolysis	Ch. 17-1

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Tentative Lecture Schedule**

#	<u>Lecture</u>	<u>Topics</u>	<u>Chapters</u> (N = notes)
	Nov 16 th , 2026	• Buffer time / optional review session for Midterm #2	N/A
	<u>Nov 18th, 2026</u>	• Midterm #2 on material from Oct 19th to Nov 9th (Test will be designed to take 60 minutes to complete, however 80 minutes will be allocated)	Ch. 14, 15, 16 and <u>N</u>
16	Nov 23 rd , 2026	• Electrogravimetric analysis • Coulometry • Amperometry	Ch. 17-2 Ch. 17-3 Ch. 17-4
17	Nov 25 th , 2026	• Voltammetry	Ch. 17-5
	Nov 30 th , 2026	• NO CLASS	N/A
	Dec 2 nd , 2026	• NO CLASS	N/A

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Tentative Lecture Schedule**

#	<u>Lecture</u>	<u>Topics</u>	<u>Chapters</u> (<u>N</u> = notes)
18	Dec 7 th , 2026	• Buffers • Diprotic Buffers • Isoelectric and isoionic pH	Ch. 9-5 Ch. 10-2 Ch. 10-6
19	Dec 9 th , 2026	• EDTA Titrations	Ch. 12
	<i>Dec 11th, 2026</i>	<i>• Buffer time / optional review session for Final Exam</i>	<i>N/A</i>
	<u>December</u> <u>2026</u> <u><i>Time: TBD</i></u>	<u>• FINAL EXAM</u> • The in-person exam will cover all of the material covered in this course and will be 3 hours in duration	Ch. 0, 28-2, 28-3, 23-2, 21, 24, 5, 6, 7, 8, 9-5, 10-2, 10-6, 12, 14, 15, 16, 17 and <u>N</u>

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